

***N*-(1-Phenylethyl)indole-2-carboxamide dimethylformamide solvate**

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Key indicators

Single-crystal X-ray study

T = 294 K

Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$

R factor = 0.054

wR factor = 0.109

Data-to-parameter ratio = 18.8

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O} \cdot \text{C}_3\text{H}_7\text{NO}$, containing indolyl and amide groups, the indolyl and phenyl rings are not coplanar, and form a dihedral angle of $88.4(2)^\circ$. There are two intermolecular hydrogen bonds through the indole N and carbonyl O atom of the molecule, as well as through the amide N atom of the molecule and carbonyl O atom of the dimethylformamide solvent molecule.

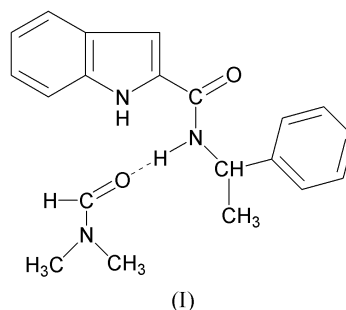
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Comment

We have reported the structure of the bidentate ligand (*R*)-*N*-(1-phenylethyl)quinoline-2-carboxamide (Yang *et al.*, 2001). This ligand was found to be easily coordinated to ruthenium(III) in a basic medium. The resulting Ru complex showed excellent catalytic activity in the epoxidation of cyclic alkenes. In order to further investigate the ligand performance of a new bidentate ligand, (I), containing the indole-2-carboxamide moiety, was synthesized. When the ligand coordinates to a metal ion through the ring N and the amide N atoms, the indolyl ligand, which contains a five-membered ring, will be less sterically hindered than the quinolyl derivative, which contains a six-membered ring. As a result, their catalytic activities should differ.



The indolyl and phenyl rings in (I) (Fig. 1) are not coplanar, and form a dihedral angle of $88.4(2)^\circ$. The amide group is rotated out of the indole ring plane, forming a dihedral angle of $2.2(1)^\circ$. There are two intermolecular hydrogen bonds; one involves the indole N and carbonyl O atom of the molecule, the other the amide N atom of the molecule and carbonyl O atom of the dimethylformamide (DMF) solvent molecule.

Experimental

The title compound was synthesized from 2-indolecarboxylic acid and methylbenzylamine, according to the general procedure of Johnson *et al.* (1960). The crystal used for the data collection was obtained by slow evaporation from a saturated DMF–water (10:1) solution at room temperature.

Crystal data

$C_{17}H_{16}N_2O \cdot C_3H_7NO$
 $M_r = 337.41$
 Triclinic, $P\bar{1}$
 $a = 7.8166$ (11) Å
 $b = 11.1683$ (18) Å
 $c = 11.5769$ (18) Å
 $\alpha = 73.310$ (3)°
 $\beta = 85.990$ (4)°
 $\gamma = 77.571$ (4)°
 $V = 945.3$ (3) Å³

$Z = 2$
 $D_x = 1.185$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 3267 reflections
 $\theta = 1-27.5^\circ$
 $\mu = 0.08$ mm⁻¹
 $T = 294$ (2) K
 Prism, colorless
 0.14 × 0.12 × 0.10 mm

Data collection

Bruker CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{min} = 0.989$, $T_{max} = 0.992$
 6524 measured reflections

4309 independent reflections
 1395 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.032$
 $\theta_{max} = 27.6^\circ$
 $h = -8 \rightarrow 10$
 $k = -14 \rightarrow 12$
 $l = -12 \rightarrow 15$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.109$
 $S = 0.97$
 4309 reflections
 229 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.33$ e Å⁻³
 $\Delta\rho_{min} = -0.15$ e Å⁻³

Table 1

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1A \cdots O1^i$	0.86	2.01	2.803 (2)	153
$N2-H2A \cdots O2$	0.86	2.06	2.863 (3)	155

Symmetry code: (i) $1 - x, -y, 1 - z$.

H atoms were placed in geometrically calculated positions and included in the final refinement in the riding-model approximation.

Data collection: SMART (Siemens, 1995); cell refinement: SMART; data reduction: SHELXTL (Siemens, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular

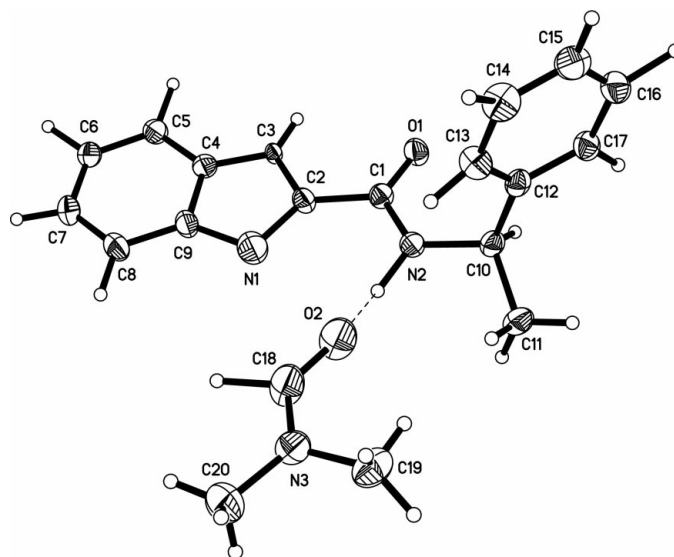


Figure 1

The molecular structure of (I), with ellipsoids at the 30% probability level (Siemens, 1995).

graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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